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(54) Title: NEGATIVE WORKING IMAGEABLE COMPOSITION CONTAINING SULFONIC ACID

(57) Abstract: The present invention provides a thermally imageable composition, which includes an acid curable composition, an acid generator, a strong acid and optionally an infrared absorber. The present invention further provides an imageable element, which includes a substrate and a thermally imageable composition according to the present invention coated on a surface of the substrate. Also provided are methods of producing an imaged element according to the present invention.

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NEGATIVE WORKING IMAGEABLE COMPOSITION CONTAINING SULFONIC ACID

1. FIELD OF THE INVENTION

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The present invention relates to an imageable composition. More particularly, the present invention relates to a thermally imageable composition having an acid curable composition, an acid generator, a strong acid and optionally an infrared absorber. As the strong acid, sulfonic acids are added to increase the speed of the thermally imageable, pre-heated, negatively working patterning compositions, which are useful especially as printing plates.

2. <u>DESCRIPTION OF THE PRIOR ART</u>

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Printing plates having a coating of a thermally imageable composition that includes an acid curable composition and either a "free acid" cure catalyst or a "latent acid" cure catalyst, such as, an acid generator, are known. However, each of these systems suffers from disadvantages, which include difficulty of controlling the cure rates.

If a "free acid" cure catalyst is used in an amount sufficient to produce a satisfactory cure speed, the shelf life of the curable composition becomes severely limited because the curing process begins immediately upon contact of the free acid with the ingredients of the acid curable composition. If, on the other hand, lesser amounts of the cure catalyst are employed to improve the shelf life, the cure rate is reduced to unacceptably low levels.

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In an alternative approach, if a "latent acid" cure catalyst is used in an amount sufficient to produce a satisfactory cure speed, the amount of WO 03/019293 PO

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energy required to generate the acid from the acid generator compound is unacceptably high even though the shelf life of the curable composition is satisfactory. If, on the other hand, lesser amounts of the "latent" cure catalyst were employed to improve the energy requirements of the acid generator, the cure rate would be reduced to unacceptably low levels.

- U.S. Patent No. 5,965,319 describes a negatively working composition having an IR absorber, novolak and resole resins, and an iodonium, sulfonium or diazonium salt, with a sulfonate group as counterion. There is no teaching that the presence of sulfonic acid as a discrete additive will improve cure rate processing latitude.
- U.S. Patent No. 6,042,987 describes a thermal negative system in which an acid is produced from a typical generator upon exposure to IR radiation. This promotes cross-linking between a novolak and a crosslinking species.

Japanese Patent Application JP 11-268,438 describes a thermal positive plate having a novolak resin and an IR absorber that has a thermally decomposable sulfonate directly attached thereto. At unimaged regions of the plate, the absorber acts as an insolubiliser and, at imaged areas, it acts as a dissolution accelerant. Japanese Patent Application JP 10-193,554 describes a negative plate having excellent run length by virtue of a new, improved polymer. Japanese Patent Application JP 3-291,665 discloses conventionally imaged negative plate systems.

U.S. Patents Nos. 5,340,699 and 5,919,601 describe imageable compositions having a binder, a crosslinker, an acid generator and an infrared absorber. These patents do not describe the use of a strong acid in addition to the acid generator or that added sulfonic acids would improve plate speed or processing latitude.

U.S. Patent No. 5,641,608 describes ablative and positive/negative working solubility differential systems for PCB precursors. The negative systems rely on adding an amine active ingredient just prior to use, or by employing a "UV flood then IR exposure" process. U.S. Patent No. 5,763,134 describes a composition having an acid generator, typically a triazine, and a squarylium dye having a defined nucleus. European Patent Application EP 632,003 describes MeO- and Me- containing phenol compounds for use as improved heat curing additives in conventional positive plates.

None of the above disclosures teach or suggest that the presence of a strong acid, such as, a sulfonic acid, will improve cure rate processing latitude when included in the composition as a discrete additive.

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Accordingly, it is an object of the present invention to improve the speed and processing robustness of thermal, pre-heated, negatively working patterning compositions, especially printing plates, while maintaining adequate shelf life.

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The present invention provides such an imageable composition, which has an improved cure rate, long shelf life of the acid curable composition and moderate energy requirement of the acid generation step.

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SUMMARY OF THE INVENTION

The present invention provides an imageable composition, which includes an acid curable composition, an acid generator and a strong acid.

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The present invention further provides an imageable element, which includes a substrate and an imageable composition coated on a surface of the substrate. The composition includes an acid curable composition, an acid generator and a strong acid.

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The present invention further provides methods of producing an imaged element.

The first of these methods, includes the steps of:

providing a thermally imageable element including a substrate and a thermally imageable composition coated on a surface of the substrate, the composition including an acid curable composition, an acid generator and a strong acid;

imagewise exposing the imageable element to heat with a hot stylus to produce an imagewise exposed element having exposed and unexposed regions;

baking the imagewise exposed element at a temperature and period of time sufficient to produce a cured element; and

contacting the cured element and a developer to remove the unexposed regions and thereby produce the imaged element.

The second of these methods, includes the steps of:

providing an imageable element including a substrate and
an imageable composition coated on a surface of the substrate, the

a strong acid;

imagewise exposing the imageable element to radiation to produce an imagewise exposed element having exposed and unexposed regions;

composition including an acid curable composition, an acid generator and

baking the imagewise exposed element at a temperature and period of time sufficient to produce a cured element; and

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contacting the cured element and a developer to remove the unexposed regions and thereby produce the imaged element.

It was found that incorporating a small amount of a sulfonic acid into pre-heat, thermal plate increases the speed of the thermal plate and allows maximum image density at wider pre-heat temperatures. In addition, it provides significantly improved processing latitude while maintaining an adequate shelf life.

10 DETAILED DESCRIPTION OF THE INVENTION

Lithographic printing plate precursors, i.e., imageable elements, typically include a radiation imageable coating applied over a support material, such as, an aluminum substrate. If after exposure to radiation, the exposed regions of the coating become soluble and are removed in the developing process, revealing the underlying hydrophilic surface of the support, the plate is called a positive-working printing plate. Conversely, if exposed regions of the plate become insoluble in the developer and the unexposed regions are removed by the developing process, the plate is called a negative-working plate. In each instance, the regions of the radiation-sensitive layer that remain (i.e., the image areas) are ink-receptive and the regions of the hydrophilic surface revealed by the developing process accept water and repel ink.

The present invention is useful in negative-working printing plates, such as Thermal Printing Plates/830, available from Kodak Polychrome Graphics, Norwalk, CT. To improve cure rate and processing latitude of the thermally imageable layer, the present invention provides an imageable composition, which in addition to an acid generator includes as a discrete additive in the composition a strong acid, such as, a sulfonic acid. Thus, the imageable composition according to the present invention

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includes an acid curable composition, an acid generator, a strong acid and optionally an infrared absorber.

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The term "hydrocarbyl" in the context of the present invention refers to a linear, branched or cyclic alkyl, alkenyl, aryl, aralkyl or alkaryl of 1 to 22 carbon atoms, substituted derivatives thereof, wherein the substituent group is selected from halogen, hydroxy, hydrocarbyloxy, carboxyl, ester, ketone, cyano, amino, amido and nitro groups. Hydrocarbyl groups in which the carbon chain is interrupted by oxygen, nitrogen or sulfur are also included in the term "hydrocarbyl".

Preferably, the imagable composition of the present invention is thermally imageable, so that the imageable element according to the present invention is infrared radiation imageable. Thus, in the method of the present invention, the step of imagewise exposing the imageable element to radiation is preferably carried out using an infrared radiation. Upon such imaging the developer solubility of the imaged area is decreased to allow differentiation between imaged and non-imaged areas upon development.

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The acid curable composition can include a binder and a crosslinking agent for that binder. Suitable binders include polymers that have at least two, preferably more than two reactive groups, such as, hydroxy, carboxylic acid, amine, carbamate, amide, sulfonamide or imide. Preferably, the reactive group is a hydroxy group, such that, the polymer can be a polyol, a polyether polyol, a novolak resin, a hydroxyfunctional acrylic resin, a hydroxyfunctional polyester resin and combination thereof.

The crosslinking agent can be any suitable crosslinking agent known in the art and includes crosslinking agents, such as, a resole resin, an amino resin, an amido resin, an epoxy compound having at least two

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epoxide groups and the like. A combination of the aforementioned crosslinkers can also be used.

The crosslinking agent preferably is an amino resin that has at least two alkoxymethyl groups, including amino resins, such as, an alkoxymethylated melamine resin, an alkoxymethylated benzoguanamine resin, an alkoxymethylated glycoluril, an alkoxymethylated polyacrylamid, an alkoxymethylated polymethacrylamid and a combination thereof. Preferably, the alkyl group in the alkoxymethylated amino resins is derived from an alcohol of 1 to 4 carbon atoms, such as methanol, ethanol, propanol, butyl alcohol, isomers thereof and mixtures thereof. Examples of such amino resins include alkoxymethylated melamine resins having from about 2 to about 6 methoxymethyl groups.

The acid curable composition can include a self-crosslinking material, such as, a resole resin. However, in addition to the resole resin, the acid curable composition can further include a polymer having at least two reactive groups, such as, hydroxy, carboxylic acid, amine, carbamate, amide, sulfonamide or imide, to form a crosslinked network with the resole resin. Preferably, the acid curable composition includes a resole resin and a novolak resin.

The acid curable composition can further include an isocyanate crosslinker that has at least two, preferably more than two isocyanate groups. Such isocyanate crosslinkers include diisocyanates, such as, isophorone diisocyanate, methylene-bis-phenyl diisocyanate, toluene diisocyanate, hexamethylene diisocyanate, tetramethylxylylene diisocyanate, dimers thereof, adducts thereof with diols or triols, and mixtures thereof.

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The imageable composition of the present invention also includes an acid generator, which is an ultraviolet, visible or infrared radiation or heat activated compound. Upon exposure to ultraviolet, visible radiation, infrared radiation or heat, either directly, or indirectly through heat transfer from an infrared absorbing compound, the acid generator produces a free acid, which acts as a cure catalyst for the curing process. Suitable acid generators include onium salts, covalently bound sulfonate group containing compounds, hydrocarbylsulfonamido-N-hydrocarbyl sulfonate and a combination thereof. Examples of the covalently bound sulfonate group containing compounds include hydrocarbyl sulfonates, such as, methyl tosylate, ethyl tosylate, benzoin tosylate, and the like.

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When an ultraviolet radiation is used with a thermally activated acid generator, the composition can further include a photothermal converter material for converting ultraviolet energy to thermal energy. In addition, a UV/visible sensitizer selected from monomolecular or polymeric compounds containing an anthracene moiety, thioxanthone moiety or alkylaminobenzophenone moiety can also be used. However, in the case of UV-activated acid generators, the use of a photothermal converter material is not necessary.

Preferably, the acid generator is an onium salt that has a non-nucleophilic counteranion, such as, tetrafluoroborate, hexafluoroarsenate, hexafluoroantimonate, hexafluorophosphate, triflate, pentafluoroethyl sulfonate, tetrakis(pentafluorophenyl)borate, p-methylbenzene sulfonate, ethyl sulfonate, trifluoromethyl acetate and pentafluoroethyl acetate.

Examples of such onium salts include iodonium salts, sulfonium salts, hydrocarbyloxysulfonium salts, hydrocarbyloxyammonium salts, aryl diazonium salts and combinations thereof. Examples of the hydrocarbyloxy ammonium salts include the salts of N-hydrocarbyloxy

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substituted nitrogen containing heterocyclic compounds, such as, N-ethoxyisoquinolinium hexafluorophosphate. Examples of the iodonium salts include 4-octyloxyphenyl phenyliodonium hexafluoroantimonate.

Preferably, the acid generator is a monomeric or oligomeric aromatic diazonium salt. The monomeric and oligomeric diazonium salts can be any diazonium salt known in the art that is suitable for use in thermal imaging.

Suitable diphenylamine-4-diazonium salts include, for example, 4diazodiphenylamine sulfate. However, any aromatic diazonium salt can be used. Examples of such aromatic diazonium salts include: diphenyl-4diazonium chloride; 2-4-(N-(naphthyl-2-methyl)-N-propylamino)benzenediazonium sulfate; chloro-diphenyl-4-diazonium chloride; 4-(3phenylpropylamino)-benzenediazonium sulfate; 4-(N-ethyl-N-(benzyl)amino)-benzenediazonium chloride; 4-(N,N-dmethyl-amino)benzenediazonium tetrafluoroborate; 4-(N-(3-phenyl-mercapto-propyl)-Nethyl-amino)-2-chlorobenzenediazonium chloride; 4-(4-methylphenoxy)benzenediazonium sulfate; 4-(phenylmercapto)-benzenediazonium chloride; 4-phenoxybenzenediazonium chloride; 4-(benzoylamino)benzenediazonium hexafluorophosphate; methylcarbazole-3diazoniumchloride; 3-methyl-diphenyleneoxide-2-diazoniumchloride, 3methyldiphenylamine-4-diazonium sulfate, 2, 3',5-trimethoxydiphenyl-4diazonium chloride: 2.4',5-triethoxydiphenyl-4-diazonium chloride: 4-(3-(3methoxyphenyl)-propylamino)-benzenediazonium sulfate; 4-(N-ethyl-N-(4methoxybenzyl)-amino)- benzenediazonium chloride: 4-(N-(naphthyl-(2)methyl)-N-n-propylamino)methoxybenzenediazonium sulfate; 4-(N-(3phenoxypropyl)-N-methylamino)-2,5-dimethoxybenzenediazonium tetrafluoroborate; 4-(N-(3-phenylmercaptopropyl)-N-ethylamino)-2-chloro-5-methoxybenzenediazonium chloride; 4-(4-(3-methylphenoxy)-phenoxy)-2,5-dimethoxybenzenediazonium sulfate; 4-(4-methoxy-phenylmercapto)-

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2,5-diethoxybenzenediazonium chloride; 2,5-diethoxy-4-phenoxybenzenediazonium chloride; 4-(3,5-dimethoxybenzoylamino)-2,5-diethoxybenzenediazonium hexafluorophosphate; methoxycarbazole-3-diazonium chloride; 3-methoxy-diphenyleneoxide-2-diazonium chloride and methoxydiphenylamine-4-diazonium sulfate.

Diazonium salts derived from the following amines are also suitable for use in the present invention: 4-amino-3-methoxydiphenylamine, 4amino-2-methoxydiphenylamine, 4'-amino-2-methoxydiphenylamine, 4'amino-4-methoxydiphenylamine, 4-amino-3-ethoxydiphenylamine, 4amino-3-hexyloxydiphenylamine, 4-amino-3-beta-hydroxyethoxydiphenylamine, 4'-amino-2-methoxy-5-methyldiphenylamine, 4-amino-3methoxy-6-methyldiphenylamine, 4'-amino-4-n-butoxydiphenylamine, 4'amino-3',4-dimethoxydiphenylamine, 4-amino-diphenylamine, 4-amino-3methyl-diphenylamine, 4-amino-3-ethyldiphenylamine, 4'-amino-3-methyldiphenylamine, 4'-amino-4-methyl-diphenylamine, 4'-amino-3,3'dimethyldiphenylamine, 3'-chloro-4-amino-diphenylamine, 4aminodiphenylamine-2-sulfonic acid, 4-aminodiphenylamine-2-carboxylic acid, 4-aminodiphenylamine-2'-carboxylic acid and 4'-bromo-4aminodiphenylamine. Preferred are 4-amino-diphenylamine, 3-methyl-4aminodiphenylamine, 3-alkoxy-4-aminodiphenylamines having 1 to 3 carbon atoms in the alkoxy group and 3-methoxy-4-aminodiphenylamine.

Preferably, the counter anion of the aromatic diazonium salt can be mesitylene sulfonate, toluene sulfonate, methane sulfonate, naphthalene sulfonate, trifluoromethane sulfonate, hexafluorophosphate and tetrafluoroborate.

Examples of the particularly preferred monomeric aromatic diazonium salts include 2-methoxy-4-phenylaminobenzenediazonium hexafluorophosphate (diazo MSPF6) represented by the formula:

2-methoxy-4-phenylaminobenzenediazonium p-toluenesulfonate represented by the formula:

$$\begin{array}{c|c} N \equiv N^+ & SO_3 \\ \hline \\ NH & CH_3 \\ \hline \end{array}$$

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and a combination thereof.

Examples of the particularly preferred oligomeric aromatic diazonium salts include compounds represented by the formula:

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which is manufactured by St. Jean Photochemicals, Quebec, Canada, under the trade name DTS-18;

$$\begin{array}{c|c}
 & \text{N} \equiv \text{N}^{+} \\
 & \text{N} \neq \text{N$$

and

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$$N \equiv N^+$$
 CH_3O
 $N \equiv N^+$
 CH_2
 CH_2

wherein n is from 1 to 11; and a combination thereof. Mixtures of any of the aforementioned diazonium salts are also suitable.

In addition to the acid generator, the imageable composition of the present invention includes an additional acid, which is a strong acid, typically introduced into the composition as the free acid. The term "strong acid" is defined herein as an acid that has a pKa of not more than about 8. Preferably, the strong acid of the present invention has a pKa of not more than about 5. More preferably, the strong acid of the present invention

has a pKa of not more than about 4. Examples of such strong acids include sulfonic acids represented by the formula:

R-SO₃H

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wherein R is a substituted or unsubstituted hydrocarbyl of 1 to 22 carbon atoms, a substituted or unsubstituted aryl of 6 to 22 carbon atoms. Mixtures of these acids can also be used to obtain desired cure rates and properties.

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The alkyl sulfonic acids can be represented by the above formula, wherein the R group is preferably a linear, branched or cyclic alkyl of 1 to 22 carbon atom or a linear, branched or cyclic haloalkyl of 1 to 22 carbon atom having at least one halogen. Mixtures of these acids can also be used. Preferably, the haloalkyl group has two or more halogen atoms. Preferred halogens include chlorine and fluorine.

The aryl sulfonic acids can be represented by the formula:

$$R^3$$
 R^4
 R^5
 R^1
 R^5

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wherein each of R¹, R², R³, R⁴ and R⁵ can independently be hydrogen, an alkyl of 1 to 12 carbon atoms, a haloalkyl of 1 to 22 carbon atoms having at least one halogen, an aryl of 6 to 12 carbon atoms, a halogen, a hydroxy, an alkoxy, a cyano, a nitro, an alkoxycarbonyl or an acyl group represented by the formula:

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wherein R⁶ can be hydrogen, alkyl of 1 to 12 carbon atoms, haloalkyl of 1 to 12 carbon atoms having at least one halogen atom, alkoxy, cyano, nitro, alkoxycarbonyl and acetyl.

Preferred aryl sulfonic acids can be represented by the formula:

$$R^6$$
 C
 R^1
 HO
 SO_3H

wherein each of R¹, R⁴ and R⁶ can be hydrogen, alkyl of 1 to 12 carbon atoms, haloalkyl of 1 to 12 carbon atoms having at least one halogen, aryl of 6 to 12 carbon atoms, halogen, hydroxy, alkoxy, cyano, nitro, alkoxycarbonyl or acyl and wherein R⁷ can be hydrogen, alkyl of 1 to 12 carbon atoms, haloalkyl of 1 to 12 carbon atoms having at least one halogen, aryl of 6 to 12 carbon atoms, alkoxycarbonyl and acyl. An example of such an aryl sulfonic acid is 3-benzoyl-4-hydroxy-6-methoxybenzenesulfonic acid.

The imageable composition of the present invention can optionally include an infrared absorber, such as, a pigment, a dye or a combination thereof.

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Examples of the preferred pigments include black pigments, yellow pigments, orange pigments, brown pigments, red pigments, purple pigments, blue pigments, green pigments, fluorescent pigments, metal powder pigments, polymer bond pigments, insoluble azo pigments, azo lake pigments, condensation azo pigments, chelate azo pigment, phthalocyanine pigments, anthraquinone pigments, perylene pigments, perynone pigments, thioindigo pigments, quinacridone pigments, dioxazine pigments, isoindolinone pigments, quinophthalone pigments, colored lake pigments, azine pigments, nitroso pigments, nitro pigments, natural pigments, fluorescent pigments, inorganic pigments, carbon black, Paris Blue, Prussian Blue or any combination thereof.

Examples of the preferred dyes include cyanine dyes, squarylium dyes, pyrylium salts and nickel thiolate complexes.

A particularly useful class of infrared absorbing dyes include compounds represented by the formula:

X.

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wherein each R¹, R², R³ and R⁴ is independently selected from the group consisting of: a linear, branched or cyclic alkyl of 1 to12 carbon atoms, alkenyl of 1 to12 carbon atoms, alkoxy of 1 to12 carbon atoms in

the alkyl and aryl of 1 to 12 carbon atoms;

wherein each pair selected from the group consisting of: R¹ and R² and R³ and R⁴ may be bonded together to form a fused aromatic ring;

wherein each R⁵, R⁶, R⁷, R⁸, R⁹ and R¹⁰ is independently selected from the group consisting of: a linear, branched or cyclic alkyl of 1 to12 carbon atoms, wherein each of said alkyls can independently have a substituent:

wherein each R¹¹, R¹² and R¹³ is independently selected from the group consisting of: hydrogen halogen, a linear, branched or cyclic alkyl of 1 to 8 carbon atoms, wherein each of said alkyls can independently have a substituent; wherein R¹² may be bonded to R¹¹ or R¹³ to form a ring structure; and when m is greater than 2, a plurality of R¹² may bonded to each other to form a ring structure;

wherein each R¹⁴ and R¹⁵ is independently selected from the group consisting of: hydrogen halogen, a linear, branched or cyclic alkyl of 1 to 8 carbon atoms, wherein each of said alkyls can independently have a substituent; wherein R¹⁴ may be bonded to R¹⁵ to form a ring structure; and when m is greater than 2, a plurality of R¹⁴ may bonded to each other to form a ring structure; and

wherein m is an integer from 1 to 8; and X is a counter arlon derived from a non-volatile acid.

Specific examples of such dyes include the following compounds: (1).

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(2).

ClO₄

5 (3).

PF₆

(4).

CIO₄

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and

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(5).

These dyes have the property of generating heat upon absorbing light. Furthermore, they have an infrared absorb band in the range from 700 to 1200 nm and thus, are suitable for use in the thermally imageable compositions of the present invention.

Further examples of dyes and pigments include triarylamine dyes, thiazolium dyes, indolium dyes, oxazolium dyes, cyanine dyes, polyaniline dyes, polypyrrole dyes, polythiophene dyes, thiolene metal complex dyes, carbon black and polymeric phthalocyanine blue pigments and those that are mentioned in U.S. Patents Nos. 5,919,601; 3,218,167; and 3,884,693, the contents of which are incorporated herein by reference in their entirety as fully set forth.

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The present invention also provides an imageable element, which employs the imageable composition of the present invention. The imageable element includes a substrate and an imageable composition according to the present invention coated on a surface of the substrate.

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The substrate of the imageable element is typically an aluminum sheet. However, other materials that are commonly known to those skilled in the art can also be used.

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Suitable substrates include any sheet material conventionally used to prepare lithographic printing plates, including metals such as aluminum

sheets; paper; paper coated on one or both sides with an alpha-olefin polymer such as polyethylene; acetate films such as polyvinyl acetate and cellulose acetate film; polyvinyl acetal film; polystyrene film; polypropylene film; polyester film such as polyethylene terephthalate film; polyamide film; polyimide film; nitrocellulose film; polycarbonate film; polyvinylchloride film; composite films such as polyester, polypropylene or polystyrene film coated with polyethylene film; metalized paper or films; metal/paper laminates; Perlon gauze; plates of magnesium, zinc, copper, anodized aluminum, electrochemically roughened aluminum, steel, and the like.

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A preferred substrate for the imageable element of present invention is an aluminum sheet. The surface of the aluminum sheet may be treated with metal finishing techniques known in the art including physical roughening, electrochemical roughening, chemical roughening, anodizing, and silicate sealing and the like. If the surface is roughened, the average roughness (Ra) is preferably in the range from 0.1 to 0.8 μ m, and more preferably in the range from about 0.1 to about 0.4 μ m.

The preferred aluminum substrate is electrochemically-grained and anodized aluminum, such as commonly used for lithographic printing plates. Anodized substrates can be prepared using sulfuric acid anodization, phosphoric acid anodization or a combination thereof. Other conventional anodization methods can also be used in the preparation of the anodized substrate of the present invention. The surfaces of these materials can be subjected to a pretreatment, if necessary, using the surface treatment techniques known in the art to improve adhesion between the substrate and organic coatings.

The imageable layer can be applied using the application methods known in the art. For example, the imageable composition can be dissolved in solvents, preferably organic solvents, and applied to an

aluminum substrate, which has been preferably grained, anodized and post-treated. Graining (or roughening) can be accomplished by mechanical or electrochemical processes or by a combination of both processes.

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After proper drying, the coating weight of the imaging layer preferably is in the range of about 0.2 to about 5.0 g/m², and more preferably in the range from about 0.7 to about 2.5 g/m².

The imageable element of the present invention is suitable for use in single as well as multilayer thermally imageable elements that are useful in lithographic printing, including lithographic printing plates that can be thermally imaged by imagewise exposure with a laser or a thermal printing head. The multilayer thermally imageable element is useful as a precursor for a lithographic printing member.

The imageable layer, the imageable element can have additional layers, such as, an underlying layer.

Possible functions of an underlying layer include:

- (1) to enhance developability of the imagewise unexposed areas; and
- (2) to act as a thermal insulating layer for the imagewise exposed areas.

Such thermal insulating polymeric layer prevents otherwise rapid heat dissipation, for example, through the heat conducting aluminum substrate. This allows more efficient thermal imaging throughout of the thermally imageable layer, particularly in the lower sections. In accordance with these functions, the underlying layer should be soluble or

at least dispersible in the developer and, preferably, have a relatively low thermal conductivity coefficient.

The thermally imageable element can further have an overlying layer. Possible functions of an overlying layer include:

- (1) to prevent damage, such as scratching, of the surface layer during handling prior to imagewise exposure; and
- (2) to prevent damage to the surface of the imagewise exposed areas, for example, by over-exposure, which could result in partial ablation.

The overlying layer should be soluble, dispersible or at least permeable to the developer.

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In a multilayer imageable case, the element has a bottom layer including a polymeric material which can be removed by treatment with an aqueous alkaline solution, and a top layer having the imageable layer of the present invention which upon imaging exhibits a decreased solubility of the exposed regions in an aqueous developer solution, i.e., is negative working.

Further, known plasticizers, adhesion promoters, flow control agents and/or UV absorbers can be added to the copying compositions of the invention. The type and quantity of such additives depend on the purpose for which the imageable element according to the present invention is intended for use. In any case, however, care must be taken that the substances added do not absorb an excessive proportion of the radiation required for acid generation and thus reduce the crosslinking sensitivity of the composition.

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Suitable plasticizers include dibutyl phthalate, diisooctyladipate, nitro esters, alkyl and aryl phosphate esters, chlorinated paraffins. Glycols or aliphatic polyols can also be added. If it is desired to ensure good storability under relative high atmospheric moisture conditions, the use of water-insoluble plasticizers is preferred.

Adhesion promoters can also be added. Suitable adhesion promoters include monomeric or polymeric organic silanes, nitrogen-containing heterocyclic compounds, such as those disclosed in U.S. Patents Nos. 3,645,722, 3,622,234, and 3,827,908, heterocyclic mercaptan compounds, mercapto alkanoic acid anilides and mercapto alkanoic acid esters.

Depending on the manner of exposure of the imageable element to radiation, the thermally imageable composition can optionally include an infrared absorber in addition to the acid curable composition, the acid generator and the strong acid. Thus, when the imagewise exposure of the thermally imageable composition to heat is carried out directly through the physical contact a hot stylus with the thermally imageable composition, the use of an infrared absorber is not necessary. On the other hand, if such imagewise exposure of the thermally imageable composition is carried out by exposing the thermally imageable composition to radiation, such as, an infrared radiation, the use of an infrared absorber is necessary.

Accordingly, the present invention provides:

(1) a first method of producing an imaged element, which includes the steps of:

providing a thermally imageable element including a substrate and

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a thermally imageable composition coated on a surface of the substrate, the composition including an acid curable composition, an acid generator and a strong acid;

imagewise exposing the imageable element to heat with a hot stylus to produce an imagewise exposed element having exposed and unexposed regions;

baking the imagewise exposed element at a temperature and period of time sufficient to produce a cured element; and contacting the cured element and a developer to remove the

unexposed regions and thereby produce the imaged element; and

(2) a second method of producing an imaged element, which includes the steps of:

providing an imageable element including a substrate and an imageable composition coated on a surface of the substrate, the composition including an acid curable composition, an acid generator and a strong acid;

imagewise exposing the imageable element to radiation to produce an imagewise exposed element having exposed and unexposed regions;

baking the imagewise exposed element at a temperature and period of time sufficient to produce a cured element; and

contacting the cured element and a developer to remove the unexposed regions and thereby produce the imaged element.

The second method can be practiced by imagewise exposing the imageable element to ultraviolet radiation provided that the thermally imageable composition includes a photothermal converting material. However, the exposing step of this method is preferably carried out using an infrared laser. Other methods such as visible or UV laser imaging may also be used, provided that a photoconverter, i.e., a photothermal converter, is present. Thus, for exposure with such visible or UV radiation

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sources, the imageable composition generally includes a photothermal converting material.

The printing plates, forms, screens, resists and the like, are prepared in the customary manner from the appropriate materials. After exposure, the non-image areas of the layer, which have retained their solubility, are removed by treatment with a suitable developer, such as, an aqueous acid or base solution.

The imaging layer of the imageable element is negative working. Preferably, the imaging layer is thermally imageable, so that the imageable element according to the present invention is infrared radiation imageable. Thus, in the method of the present invention, the step of imagewise exposing the imageable element to radiation is carried out using an infrared radiation. Upon such imaging the developer solubility of the imaged area is decreased to allow differentiation between imaged and non-imaged areas upon development.

Following imagewise exposure by analog or digital means, an imaged element having exposed areas and complimentary unexposed areas is obtained. Thereafter, the exposed plate precursor is baked at a temperature from about 220 °F to about 280 °F, preferably from about 240 °F to about 260 °F for a period of time from about 45 seconds to about 75 seconds, preferably from about 55 seconds to about 65 seconds. The exposed plate precursor is then developed with a developer to selectively remove the uncrosslinked materials in the unexposed regions.

The developer composition is dependent on the nature of the polymeric substance, but is preferably an aqueous composition. Common components of aqueous developers include surfactants, chelating agents, such as, salts of ethylenediamine tetraacetic acid, organic solvents, such

as, benzyl alcohol, ethylene glycol monomethyl ether, ethylene glycol monoethyl ether, diacetone alcohol, butyl acetate, ethylene glycol methyl ether acetate, methyl isobutyl ketone and a mixture thereof, and alkaline components, such as organic amines, including alkanol amines, such as triethanol amine and methyl diethanol amine, inorganic metasilicates, organic metasilicates, hydroxides and bicarbonates.

The pH of the aqueous developer is preferably within about 5 to about 14, depending on the nature of the composition of the imaging layer. The development can be performed by any known manner, for instance, by rubbing the plate surface with a developing pad containing the foregoing developer or by pouring the developer on the plate surface and then rubbing the surface with a developing brush in water.

The inventors have surprisingly discovered that incorporating a small amount of a sulfonic acid into pre-heat, thermal plate increases the speed of the thermal plate and allows maximum image density at wider pre-heat temperatures. In addition, it provides significantly improved processing latitude while maintaining an adequate shelf life.

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For example, when an experiment was completed adding 0 to 1 % 3-benzoyl-4-hydroxy-6-methoxybenzenesulfonic acid (HMBS) into a series of coatings for thermal pre heat plates, dramatic increases in speed were found at 0.04% level. In addition the formulation had a 30 to 40 °F pre heat window. With pre heat plates including the 3-benzoyl-4-hydroxy-6-methoxybenzene sulfonic acid (HMBS), the plate speed was around 80 mJ/cm⁻² with a 250 °F pre-heat temperature, when the plate system was developed in MX1710.

The present invention further provides an improved formulating scope. Previous attempts to improve speed and fog control have had

deleterious effects on plate performance. Thus, if one is able to prepare a fast plate, a plate having a poor shelf life is obtained. Conversely, if a plate having a good shelf life is prepared, a slow plate is obtained. In contrast, the present invention provides a substantial improvement in the speed and processing robustness of thermal, pre-heated, negatively working patterning compositions, such as, printing plates, forms, screens, resists, and the like, while maintaining adequate shelf life, improvements in plate speed and increased processing latitude of thermal, pre-heat negative plates.

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The invention is further described in the following examples, which are intended to be illustrative and not limiting.

EXAMPLES

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The following are referred to hereinafter:

- (1) Resin A a resole resin, UCAR phenolic resin BKS-5928 as supplied by Union Carbide Corporation.
- (2) Resin B a novolak resin, N-9P as supplied by Eastman Kodak Company.
 - (3) Acid generator A Diazo MSPF6 as supplied by Diversitec Corporation, Fort Collins, Colorado having the following structure:

25 (4) IR dye A – having the following structure:

- (5) Blue dye A Simbase Victoria Blue FBO 23363, as supplied by Charkit Chemical Corporation, Darien, Connecticut.
- (6) Terephthaldehyde as supplied by Aldrich Chemical5 Company, Milwaukee, WI.
 - (7) XDSA, having the following structure:

- (8) HMBS 3-benzoyl-4-hydroxy-6-methoxybenzenesulfonic acid as supplied by Aldrich.
- (9) Substrate 0.3 mm thick sheets of aluminum that had been electrograined, anodised and post-anodically treated with an aqueous solution of an inorganic phosphate.
- (10) Developer A MX1710 as supplied by Kodak Polychrome Graphics.

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- (11) Creo trendsetter 3244: a commercially available platesetter, operating at a wavelength of 830 nm, as supplied by Creo products of Canada.
- (12) Gretag Macbeth D19C densitometer: a commercially available densitometer as supplied by Color data systems Limited of the Wirral, UK.

Examples C1, 1 to 4:

Coating formulations included solutions in 1-methoxypropan-2-ol / acetone 97:3 (w:w) of the components described in Table 1. Plates were prepared by coating the formulations onto the aluminum substrate by means of a wire wound bar. The formulation concentrations were selected to provide dry films having a coating weight of 120 mg/ft. The film weights were measured after thorough air drying over several days.

Table 1

	Example					
	C1	1	2	3	4	
Component	Parts by Weight					
Resin A	39.99	39.97	39.89	39.79	39.59	
Resin B	51.23	51.20	51.11	50.97	50.71	
Acid Generator A	2.57	2.57	2.56	2.56	2.55	
IR Dye A	2.77	2.77	2.76	2.75	2.74	
Terephthaldehyde	2.28	2.28	2.28	2.27	2.26	
Blue Dye A	0.30	0.30	0.30	0.30	0.30	
XDSA	0.86	0.86	0.86	0.86	0.85	
HMBS a	1	0.05	0.24	0.50	1.00	

^a HMBS: 3-benzoyl-4-hydroxy-6-methoxybenzenesulfonic acid

The plates were then imaged on the Creo Trendsetter at 76, 81, 86, 93, 100, 109 and 119 mJ/cm⁻², using an internal solid image pattern (100% exposure). After imaging the plates were heated in a forced air oven at 250 °F for 1 minute, cooled and processed in a mechanical processor using developer A. The solid density (D_{max}) of the remaining coating was then measured using the Gretag densitometer. The results are shown in Table 2 below.

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Table 2

	D _{max} at stated imaging energy density (mJ/cm ⁻²)						
	76	81	86	93	100	109	119
Example							
C1	0.55	0.56	0.72	0.96	1.06	1.27	1.39
1	0.58	0.66	0.91	1.10	1.24	1.40	1.41
2	1.08	1.26	1.39	1.41	1.41	1.41	1.41
3	1.27	1.34	1.41	1.39	1.41	1.41	1.41
4	1.31	1.35	1.39	1.39	1.39	1.39	1.39

Table 2 shows that addition of sulfonic acid reduces the amount of imaging energy required to achieve an exposed negative working coating.

Further, plate samples of examples C1 and 1 to 4 were aged in an environmental cabinet (80% RH, 104 F) for 5 and 10 days prior to imaging, pre-heating and processing. After processing these aged plates were also read via the Gretag densitometer.

The results are summarized in Table 3.

Table 3

	D _{max} at stated imaging energy density (mJ/cm ⁻²)						
	76	81	86	93	100	109	119
Example							
Aged 5 days				,			
C1	0.84	1.10	1.32	1.42	1.45	1.44	1.44
1	1.08	1.29	1.37	1.46	1.46	1.48	1.46
2	1.15	1.36	1.50	1.53	1.53	1.52	1.54
3	1.45	1.51	1.52	1.53	1.53	1.52	1.52
Aged 10 days							-
C1	0.77	1.11	1.39	1.47	1.51	1.51	1.51
1	1.08	1.36	1.49	1.51	1.51	1.55	1.56
2	1.16	1.35	1.42	1.43	1.51	1.51	1.53
3	1.47	1.54	1.52	1.50	1.53	1.52	1.50
4	1.36	1.49	1.54	1.55	1.53	1.53	1.53

Table 3 shows that the presence of the sulfonic acid does not affect the characteristic aging of the plates. The addition of sulfonic acid continues to reduce the amount of imaging energy required to achieve an exposed negative working coating with aged plates.

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The effect of the sulfonic acid on the pre-heating latitude of the plate system, was investigated using plate samples from examples C1 and 1. The plates were imaged on the Creo Trendsetter as described above. After imaging, the plates were heated in a forced air oven at 245, 250, 255, 260, 265 and 270 F for 1 minute, cooled and processed as previously

described. The solid density (D_{max}) of the remaining coating was then measured using the Gretag densitometer.

The results can be seen in Table 4 below.

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Table 4

	D _{max} at stated imaging energy density (mJ/cm ⁻²)						
	76	81	86	93	100	109	119
Pre-heat = 245 F							
C1	0.61	0.62	0.63	0.69	0.89	1.07	1.33
1	0.68	0.84	1.11	1.19	1.35	1.39	1.45
Pre-heat = 250 F							
C1	0.61	0.63	0.65	0.65	1.04	1.34	1.50
1	0.75	1.13	1.23	1.37	1.49	1.55	1.57
Pre-heat = 255 F							
C1	0.64	0.69	0.81	0.95	1.19	1.39	1.47
1	1.08	1.32	1.44	1.51	1.54	1.55	1.55
Pre-heat = 260 F							
C1	0.75	0.9	1.05	1.18	1.34	1.43	1.47
1	1.16	1.31	1.45	1.48	1.52	1.54	1.56
Pre-heat = 265 F							
C1	1.07	1.17	1.33	1.35	1.41	1.43	1.45
1	1.44	1.47	1.53	1.55	1.55	1.57	1.58
Pre-heat = 270 F							
C1	1.36	1.42	1.47	1.52	1.54	1.57	1.56
1	1.44	1.48	1.50	1.51	1.53	1.53	1.53

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Table 4 shows that for a given imaging energy density, the presence of the sulfonic acid allows maximum image density at lower preheat temperatures.

The present invention has been described with particular reference to the preferred embodiments. It should be understood that variations and modifications thereof can be devised by those skilled in the art without departing from the spirit and scope of the present invention. Accordingly, the present invention embraces all such alternatives, modifications and variations that fall within the scope of the appended claims.

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WHAT IS CLAIMED IS:

- An imageable composition comprising:

 an acid curable composition;
 an acid generator; and
 a strong acid.
- 2. The composition of claim 1, wherein said acid curable composition comprises:
 - a binder; and
 - a crosslinking agent for said binder.
- 3. The composition of claim 2, wherein said binder comprises a polymer having at least two reactive groups each independently selected from the group consisting of: hydroxy, carboxylic acid, amine, carbamate, amide, sulfonamide and imide.
- 4. The composition of claim 3, wherein said reactive group in said polymer is a hydroxy group.
 - 5. The composition of claim 4, wherein said polymer is selected from the group consisting of: a polyol, a polyether polyol, a novolak resin, a resole resin, a hydroxyfunctional acrylic resin, a hydroxyfunctional polyester resin and combination thereof.
 - 6. The composition of claim 2, wherein said binder is a novolak resin.
- 7. The composition of claim 2, wherein said crosslinking agent is selected from the group consisting of: a resole resin, an amino resin, an

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amido resin, an epoxy compound having at least two epoxide groups and a combination thereof.

- 8. The composition of claim 7, wherein said crosslinking agent is resole resin.
 - 9. The composition of claim 8, wherein said binder is a novolak resin.
- 10. The composition of claim 7, wherein said crosslinking agent is an amino resin having at least two alkoxymethyl groups.
 - 11. The composition of claim 10, wherein said amino resin is selected from the group consisting of: an alkoxymethylated melamine resin, an alkoxymethylated benzoguanamine resin, an alkoxymethylated glycoluril, an alkoxymethylated polyacrylamid, an alkoxymethylated polymethacrylamid and a combination thereof.
- 12. The composition of claim 11, wherein said amino resin is an alkoxymethylated melamine resin having from about 2 to about 6 methoxymethyl groups.
 - 13. The composition of claim 2, further comprising an isocyanate crosslinker having at least two isocyanate groups.
 - 14. The composition of claim 1, wherein said acid generator is an ultraviolet, visible or infrared radiation or heat activated compound.
- 15. The composition of claim 14, wherein said an acid generator is selected from the group consisting of: an onium salt, a covalently bound

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sulfonate group containing compound, hydrocarbylsulfonamido-N-hydrocarbyl sulfonate and a combination thereof.

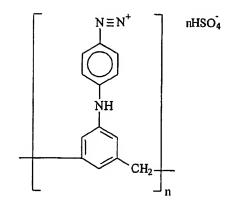
- 16. The composition of claim 15, wherein said acid generator is an onium salt.
 - 17. The composition of claim 16, wherein said onium salt has a non-nucleophilic counteranion selected from the group consisting of: tetrafluoroborate, hexafluorophosphate, hexafluoroarsenate, hexafluoroantimonate, triflate, tetrakis(pentafluorophenyl)borate, pentafluoroethyl sulfonate, p-methylbenzene sulfonate, ethyl sulfonate, trifluoromethyl acetate and pentafluoroethyl acetate.
- 18. The composition of claim 16, wherein said onium salt is selected from the group consisting of: an iodonium salt, a sulfonium salt, a hydrocarbyloxysulfonium salt, a hydrocarbyloxyammonium salt, an aryl diazonium salt and a combination thereof.
- 19. The composition of claim 18 wherein said
 hydrocarbyloxyammonium salt is a salt of an N-hydrocarbyloxy substituted
 nitrogen containing heterocyclic compound.
 - 20. The composition of claim 19 wherein said N-hydrocarbyloxy substituted nitrogen containing heterocyclic compound is N-ethoxyisoquinolinium hexafluorophosphate.
 - 21. The composition of claim 18, wherein said iodonium salt is 4-octyloxyphenyl phenyliodonium hexafluoroantimonate.
 - 22. The composition of claim 18, wherein said acid generator is a monomeric or oligomeric aromatic diazonium salt.

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- 23. The composition of claim 22, wherein said diazonium salt is selected from the group consisting of: 2-methoxy-4-phenylaminobenzene diazonium hexafluorophosphate (diazo MSPF6) represented by the formula:
 - CH₃O N≡N⁺ PF₆

2-methoxy-4-phenylaminobenzenediazonium p-toluenesulfonate represented by the formula:

an oligomeric diazonium salt selected from the group consisting of compounds represented by the formula:



$$N \equiv N^{+}$$
 $N \equiv N^{+}$
 $N \equiv$

5 and

$$\begin{array}{c|c}
 & \text{N} \equiv \text{N}^{+} \\
 & \text{CH}_{3}\text{O} \\
 & \text{NH} \\
 & \text{CH}_{2} \\
 & \text{n}
\end{array}$$

wherein n is from 1 to 11; and a combination of any of the aforementioned compounds.

- 24. The composition of claim 1, wherein said strong acid is an acid having a pKa of not more than about 8.
 - 25. The composition of claim 1, wherein said strong acid is an acid having a pKa of not more than about 4.
- 10 26. The composition of claim 24, wherein said strong acid is a sulfonic acid represented by the formula:

R-SO₃H

- wherein R is selected from the group consisting of: a substituted or unsubstituted hydrocarbyl of 1 to 22 carbon atoms, a substituted or unsubstituted aryl of 6 to 22 carbon atoms and a mixture thereof.
- 27. The composition of claim 26, wherein R is selected from the group consisting of: linear, branched or cyclic alkyl of 1 to 22 carbon atom, linear, branched or cyclic haloalkyl of 1 to 22 carbon atom having at least one halogen and a mixture thereof.
- 28. The composition of claim 26, wherein said sulfonic acid is an aryl sulfonic acid represented by the formula:

$$R^3$$
 R^4
 R^5

wherein each of R¹, R², R³, R⁴ and R⁵ is independently selected from the group consisting of: hydrogen, alkyl of 1 to 12 carbon atoms, haloalkyl of 1 to 22 carbon atoms having at least one halogen, aryl of 6 to 12 carbon atoms, halogen, hydroxy, alkoxy, cyano, nitro, alkoxycarbonyl and acyl.

29. The composition of claim 28, wherein said acyl is represented by the formula:

wherein R⁶ is selected from the group consisting of: hydrogen, alkyl of 1 to 12 carbon atoms, haloalkyl of 1 to 12 carbon atoms having at least one halogen, alkoxy, cyano, nitro, alkoxycarbonyl and acetyl

30. The composition of claim 28, wherein said aryl sulfonic acid is represented by the formula:

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wherein each of R¹, R⁴ and R⁶ is independently selected from the group consisting of: hydrogen, alkyl of 1 to 12 carbon atoms, haloalkyl of 1 to 12 carbon atoms having at least one halogen, aryl of 6 to 12 carbon atoms, halogen, hydroxy, alkoxy, cyano, nitro, alkoxycarbonyl and acyl and wherein R⁷ is selected from the group consisting of: hydrogen, alkyl of 1 to 12 carbon atoms, haloalkyl of 1 to 12 carbon atoms having at least one halogen, aryl of 6 to 12 carbon atoms, alkoxycarbonyl and acyl.

- 31. The composition of claim 30, wherein said aryl sulfonic acid is 3-benzoyl-4-hydroxy-6-methoxybenzenesulfonic acid.
 - 32. The composition of claim 1, further comprising a photothermal converter material.
 - 33. The composition of claim 1, further comprising an infrared absorber.
 - 34. The composition of claim 33, wherein said infrared absorber is selected from the group consisting of: a pigment, a dye and a combination thereof.
 - 35. The composition of claim 34, wherein said pigment is selected from the group consisting of: black pigments, yellow pigments, orange pigments, brown pigments, red pigments, purple pigments, blue pigments, green pigments, fluorescent pigments, metal powder pigments, polymer bond pigments, insoluble azo pigments, azo lake pigments, condensation azo pigments, chelate azo pigment, phthalocyanine pigments, anthraquinone pigments, perylene pigments, perynone pigments, thioindigo pigments, quinacridone pigments, dioxazine pigments, isoindolinone pigments, quinophthalone pigments, colored lake pigments, azine pigments, nitroso pigments, nitro pigments, natural

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pigments, fluorescent pigments, inorganic pigments, carbon black, Paris Blue, Prussian Blue and a combination thereof.

- 36. The composition of claim 34, wherein said infrared absorber
 is a dye selected from the group consisting of: cyanine dyes, squarylium dyes, pyrylium salts and nickel thiolate complexes.
 - 37. An imageable element comprising:

a substrate; and

an imageable composition coated on a surface of said substrate, said composition comprising: an acid curable composition; an acid generator; and a strong acid.

- 38. The imageable element of claim 37, wherein said thermally imageable composition further comprises an infrared absorber.
 - 39. The imageable element of claim 37, wherein said thermally imageable composition comprises a photothermal converting material.
 - 40. A method of producing an imaged element comprising the steps of:

providing a thermally imageable element comprising a substrate and a thermally imageable composition coated on a surface of said substrate, said composition comprising an acid curable composition, an acid generator and a strong acid;

imagewise exposing said imageable element to heat with a hot stylus to produce an imagewise exposed element having exposed and unexposed regions;

baking said imagewise exposed element at a temperature and period of time sufficient to produce a cured element; and

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contacting said cured element and a developer to remove the unexposed regions and thereby produce said imaged element.

41. A method of producing an imaged element comprising the steps of:

providing an imageable element comprising a substrate and an imageable composition coated on a surface of said substrate, said composition comprising an acid curable composition, an acid generator and a strong acid;

imagewise exposing said imageable element to radiation to produce an imagewise exposed element having exposed and unexposed regions;

baking said imagewise exposed element at a temperature and period of time sufficient to produce a cured element; and

contacting said cured element and a developer to remove the unexposed regions and thereby produce said imaged element.

- 42. The method of claim 41, wherein said thermally imageable composition further comprises an infrared absorber
- 43. The method of claim 41, wherein said thermally imageable composition comprises a photothermal converting material.
- 44. The method of claim 41, wherein said exposing step is carried out using an infrared laser.

INTERNATIONAL SEARCH REPORT

International application No.

PCT/ US0/220822

A. CLASSIFICATION OF SUBJECT MATTER IPC(7) : GO3F 7/021, 7/038, 7/30, 7/38						
IPC(7) : GO3F 7/021, 7/038, 7/30, 7/38 US CL : 430/157, 175, 176,, 302						
According to International Patent Classification (IPC) or to both national classification and IPC						
B. FIELDS SEARCHED						
Minimum documentation searched (classification system follow U.S.: 430/157, 175, 176,, 302, 270.1	red by classification symbols)					
Documentation searched other than minimum documentation to	the extent that such documents are include	d in the fields searched				
Electronic data base consulted during the international search (I EAST, WEST, DERWENT	name of data base and, where practicable, s	search terms used)				
C. DOCUMENTS CONSIDERED TO BE RELEVANT						
Category * Citation of document, with indication, where	appropriate of the relevant passages	Relevant to claim No.				
X US 5,340,699 A (HALEY et al) 23 AUGUST 19	94 (23.08.1994), column 7 ilne 3 -	1-44				
column 8, line 4	(core of 255), Goldani /, mio 5 -	1-44				
		- .				
		-				
Further documents are listed in the continuation of Box C.	See patent family annex.	-				
* Special categories of cited documents: "A" document defining the general state of the art which is not considered to be of particular relevance	"T" later document published after the inter date and not in conflict with the applic principle or theory underlying the inve	ation but cited to understand the				
B earlier application or patent published on or after the international filling date	"X" document of particular relevance; the considered novel or cannot be consider	daimed invention cannot be ed to involve an inventive step				
"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	when the document is taken alone "Y" document of particular relevance: the c	slaimed invention cannot be				
O document referring to an oral disclosure, use, exhibition or other means	considered to involve an inventive step combined with one or more other such being obvious to a person skilled in the	documents, such combination				
"P" document published prior to the international filing date but later than the priority date claimed	"&" document member of the same patent for	amily				
Date of the actual completion of the international search Date of mailing of the international search report 18 November 2002 (18.11.2002) 18 DEC 2002						
18 November 2002 (18.11.2002) Name and mailing address of the ISA/US		J 2002				
Commissioner of Patents and Trademarks Box PCT John Cim						
Washington, D.C. 20231 Facsimile No. (703)305-3230 Telephone No. 703-308-0661						
orm PCT/ISA/210 (second sheet) (July 1998)						